Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

(*E*)-*N*′-(3-Fluorobenzylidene)-3-nitrobenzohydrazide

Xiao-Yan Li

Zibo Vocational Institute, Zibo 255314, People's Republic of China
Correspondence e-mail: lixiaoyan_zb@126.com

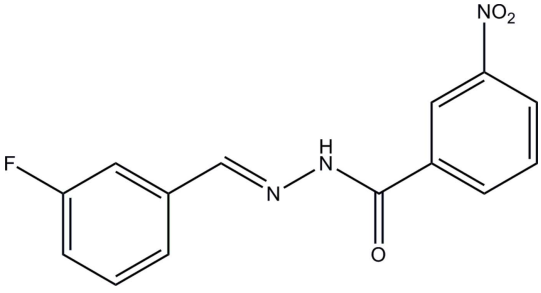
Received 2 February 2012; accepted 7 February 2012

Key indicators: single-crystal X-ray study; T = 298 K; mean σ(C–C) = 0.005 Å; R factor = 0.080; wR factor = 0.218; data-to-parameter ratio = 12.6.

In the title compound, $C_{14}H_{10}FN_3O_3$, the molecule exists in a *trans* conformation with respect to the methylidene unit. The dihedral angle between the benzene rings is 5.1 (2)°. In the crystal, molecules are linked through N—H···O hydrogen bonds, forming chains along the *c* axis.

Related literature

For the syntheses and crystal structures of hydrazone compounds, see: Hashemian *et al.* (2011); Lei (2011); Shalash *et al.* (2010). For the crystal structures of similar compounds, reported recently by the author, see: Li (2011*a*,*b*).



Experimental

Crystal data

$C_{14}H_{10}FN_3O_3$
M_r = 287.25
Monoclinic, $P2_1/c$
a = 11.823 (2) Å
b = 12.813 (3) Å
c = 8.7020 (17) Å
β = 94.855 (2)°
V = 1313.5 (5) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.11 mm^{-1}
T = 298 K
0.17 × 0.17 × 0.15 mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.981, T_{max} = 0.983
9452 measured reflections
2429 independent reflections
1438 reflections with $I > 2\sigma(I)$
R_{int} = 0.075

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.080
$wR(F^2)$ = 0.218
S = 1.02
2429 reflections
193 parameters
1 restraint
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 0.59 e Å^{-3}
$\Delta\rho_{min}$ = −0.22 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N2—H2···O3^{i}	0.90 (1)	1.96 (1)	2.846 (4)	172 (5)

Symmetry code: (i) $x, -y+\frac{1}{2}, z-\frac{1}{2}$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1998); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

The author is grateful to the Zibo Vocational Institute for supporting this work.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RZ2708).

References

Bruker (1998). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Hashemian, S., Ghaeinee, V. & Notash, B. (2011). *Acta Cryst.* E**67**, o171.
Lei, Y. (2011). *Acta Cryst.* E**67**, o162.
Li, X.-Y. (2011*a*). *Acta Cryst.* E**67**, o1798.
Li, X.-Y. (2011*b*). *Acta Cryst.* E**67**, o2511.
Shalash, M., Salhin, A., Adnan, R., Yeap, C. S. & Fun, H.-K. (2010). *Acta Cryst.* E**66**, o3126–o3127.
Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.

supplementary materials

Acta Cryst. (2012). E**68**, o697 [doi:10.1107/S1600536812005478]

(*E*)-*N*′-(3-Fluorobenzylidene)-3-nitrobenzohydrazide

Xiao-Yan Li

Comment

In recent years, hydrazone compounds have attracted much attention due to their syntheses and crystal structures (Hashemian *et al.*, 2011; Lei, 2011; Shalash *et al.*, 2010). As a continuation of our work on such compounds (Li, 2011*a*,*b*), the author reports herein on the crystal structure of the new title hydrazone compound.

The title compound (Fig. 1) exists in a *trans* configuration with respect to the methylidene unit. The dihedral angle between the C1–C6 and C9–C14 benzene rings of the molecule is 5.1 (2)°. The N1/O1/O2 nitro group is tilted by 16.3 (3)° with respect to the C1—C6 benzene ring. In the crystal, molecules are linked through N–H···O hydrogen bonds (Table 1) to form chains along the *c* axis (Fig. 2).

Experimental

A mixture of 3-fluorobenzaldehyde (0.124 g, 1 mmol) and 3-nitrobenzohydrazide (0.181 g, 1 mmol) in 30 ml of ethanol containing few drops of acetic acid was refluxed for about 1 h. On cooling to room temperature, a solid precipitate was formed. The solid was filtered and then recrystallized from methanol. Yellow crystals, suitable for X-ray diffraction analysis, were obtained by slow evaporation of the solvent.

Refinement

The amino H atom was located from a difference Fourier map and was refined isotropically with the N—H distance restrained to 0.90 (1) Å. The remaining H-atoms were positioned geometrically and refined using a riding model, with C–H = 0.93 Å, and with U_{iso}(H) set to 1.2U_{eq}(C).

Computing details

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1998); data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

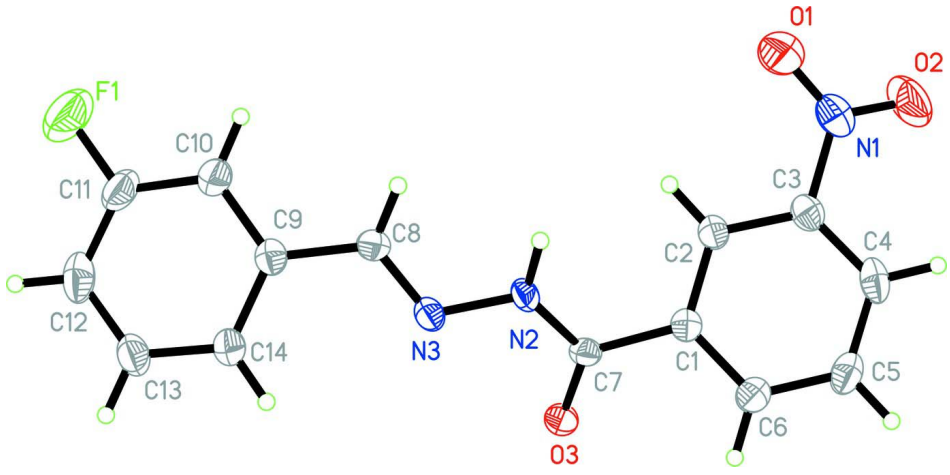


Figure 1

The molecular structure of the title compound, showing displacement ellipsoids drawn at the 30% probability level.

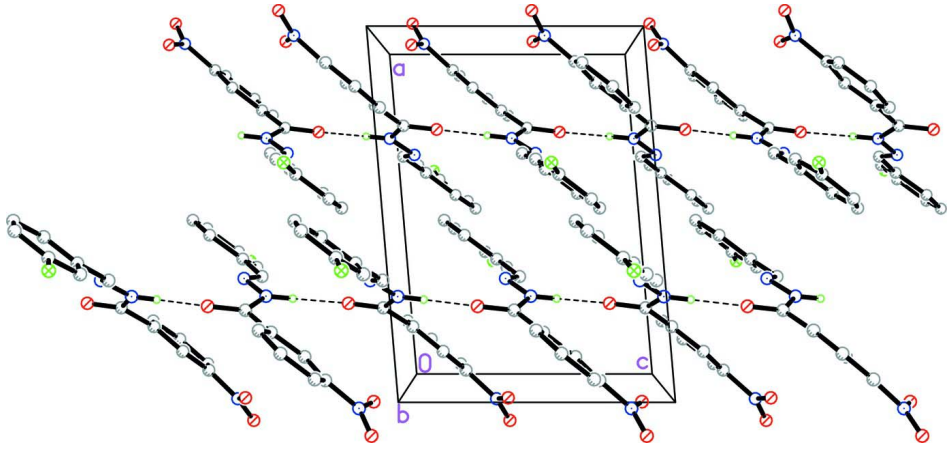


Figure 2

Molecular packing diagram of the title compound, viewed along the *b* axis. Hydrogen bonds are indicated by dashed lines. The C-bound H-atoms have been omitted for clarity.

(*E*)-*N*′-(3-Fluorobenzylidene)-3-nitrobenzohydrazide

Crystal data

$C_{14}H_{10}FN_3O_3$
$M_r = 287.25$
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
$a = 11.823$ (2) Å
$b = 12.813$ (3) Å
$c = 8.7020$ (17) Å
$\beta = 94.855$ (2)°
$V = 1313.5$ (5) Å^3
$Z = 4$

$F(000) = 592$
$D_x = 1.453$ Mg m^{-3}
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 690 reflections
$\theta = 2.4$–26.2°
$\mu = 0.11$ mm^{-1}
$T = 298$ K
Block, yellow
0.17 × 0.17 × 0.15 mm

Data collection

Bruker SMART CCD area-detector diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
ω scans
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.981, T_{max} = 0.983
9452 measured reflections
2429 independent reflections
1438 reflections with $I > 2\sigma(I)$
R_{int} = 0.075
θ_{max} = 25.5°, θ_{min} = 1.7°
h = −14→14
k = −15→15
l = −10→10

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.080
$wR(F^2)$ = 0.218
S = 1.02
2429 reflections
193 parameters
1 restraint
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H atoms treated by a mixture of independent and constrained refinement
$w = 1/[\sigma^2(F_o^2) + (0.0985P)^2 + 0.4222P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max}$ = 0.59 e Å^{-3}
$\Delta\rho_{min}$ = −0.22 e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	U_{iso}*/U_{eq}
F1	0.6421 (4)	−0.2974 (2)	1.1301 (4)	0.1160 (14)
N1	1.0252 (3)	0.4021 (4)	0.6545 (5)	0.0679 (12)
N2	0.7281 (3)	0.2345 (2)	0.9891 (3)	0.0395 (8)
N3	0.6810 (3)	0.1600 (2)	1.0796 (3)	0.0378 (8)
O1	1.0149 (4)	0.3120 (3)	0.6079 (6)	0.1118 (16)
O2	1.0930 (3)	0.4628 (3)	0.6109 (4)	0.0929 (13)
O3	0.7513 (2)	0.34723 (19)	1.1894 (3)	0.0479 (8)
C1	0.8197 (3)	0.3996 (3)	0.9543 (4)	0.0342 (9)
C2	0.8882 (3)	0.3651 (3)	0.8433 (4)	0.0375 (9)
H2A	0.8947	0.2943	0.8222	0.045*
C3	0.9466 (3)	0.4386 (3)	0.7646 (4)	0.0427 (10)
C4	0.9357 (4)	0.5431 (3)	0.7881 (5)	0.0527 (11)
H4	0.9744	0.5910	0.7318	0.063*
C5	0.8655 (4)	0.5763 (3)	0.8979 (5)	0.0538 (11)
H5	0.8568	0.6473	0.9156	0.065*
C6	0.8089 (3)	0.5051 (3)	0.9806 (5)	0.0445 (10)

H6	0.7628	0.5282	1.0551	0.053*
C7	0.7631 (3)	0.3250 (3)	1.0550 (4)	0.0354 (9)
C8	0.6781 (3)	0.0680 (3)	1.0258 (4)	0.0367 (9)
H8	0.7070	0.0547	0.9316	0.044*
C9	0.6300 (3)	−0.0173 (3)	1.1103 (4)	0.0385 (9)
C10	0.6562 (4)	−0.1197 (3)	1.0745 (5)	0.0529 (11)
H10	0.7025	−0.1344	0.9961	0.064*
C11	0.6121 (4)	−0.1984 (3)	1.1575 (5)	0.0619 (13)
C12	0.5428 (4)	−0.1806 (4)	1.2734 (5)	0.0634 (13)
H12	0.5138	−0.2358	1.3273	0.076*
C13	0.5176 (4)	−0.0806 (4)	1.3071 (5)	0.0546 (12)
H13	0.4702	−0.0673	1.3848	0.066*
C14	0.5605 (3)	0.0019 (3)	1.2291 (4)	0.0447 (10)
H14	0.5433	0.0701	1.2555	0.054*
H2	0.736 (4)	0.215 (4)	0.891 (2)	0.080*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.192 (4)	0.0477 (18)	0.111 (3)	−0.004 (2)	0.029 (3)	−0.0008 (17)
N1	0.065 (3)	0.073 (3)	0.069 (3)	−0.009 (2)	0.028 (2)	0.006 (2)
N2	0.054 (2)	0.0383 (18)	0.0278 (16)	−0.0094 (16)	0.0141 (15)	0.0014 (14)
N3	0.0390 (18)	0.0411 (19)	0.0342 (17)	−0.0043 (15)	0.0084 (13)	0.0036 (14)
O1	0.133 (4)	0.077 (3)	0.139 (4)	−0.010 (3)	0.088 (3)	−0.010 (3)
O2	0.087 (3)	0.109 (3)	0.089 (3)	−0.039 (2)	0.042 (2)	0.002 (2)
O3	0.0715 (19)	0.0427 (16)	0.0312 (14)	−0.0021 (14)	0.0154 (13)	−0.0040 (12)
C1	0.036 (2)	0.036 (2)	0.0298 (18)	−0.0029 (17)	−0.0025 (16)	−0.0001 (16)
C2	0.041 (2)	0.039 (2)	0.033 (2)	−0.0069 (18)	0.0019 (17)	0.0001 (16)
C3	0.039 (2)	0.049 (3)	0.041 (2)	−0.0064 (18)	0.0036 (17)	0.0022 (18)
C4	0.058 (3)	0.053 (3)	0.048 (2)	−0.021 (2)	0.004 (2)	0.013 (2)
C5	0.067 (3)	0.034 (2)	0.059 (3)	−0.007 (2)	0.000 (2)	0.003 (2)
C6	0.048 (2)	0.040 (2)	0.044 (2)	0.0012 (19)	0.0018 (19)	−0.0007 (18)
C7	0.040 (2)	0.037 (2)	0.031 (2)	0.0026 (17)	0.0073 (16)	−0.0030 (16)
C8	0.037 (2)	0.039 (2)	0.035 (2)	0.0036 (17)	0.0050 (16)	−0.0006 (17)
C9	0.039 (2)	0.041 (2)	0.035 (2)	−0.0062 (18)	−0.0013 (17)	0.0014 (17)
C10	0.069 (3)	0.039 (2)	0.052 (3)	−0.003 (2)	0.014 (2)	0.001 (2)
C11	0.092 (4)	0.031 (2)	0.063 (3)	−0.001 (2)	0.008 (3)	0.001 (2)
C12	0.080 (3)	0.055 (3)	0.055 (3)	−0.025 (3)	0.000 (2)	0.015 (2)
C13	0.058 (3)	0.057 (3)	0.051 (3)	−0.014 (2)	0.015 (2)	0.006 (2)
C14	0.049 (2)	0.041 (2)	0.045 (2)	−0.0040 (19)	0.0083 (19)	0.0042 (18)

Geometric parameters (Å, º)

F1—C11	1.344 (5)	C4—H4	0.9300
N1—O2	1.201 (5)	C5—C6	1.372 (6)
N1—O1	1.225 (5)	C5—H5	0.9300
N1—C3	1.467 (5)	C6—H6	0.9300
N2—C7	1.343 (5)	C8—C9	1.459 (5)
N2—N3	1.383 (4)	C8—H8	0.9300
N2—H2	0.897 (10)	C9—C10	1.390 (6)

N3—C8	1.268 (4)	C9—C14	1.396 (5)
O3—C7	1.223 (4)	C10—C11	1.369 (6)
C1—C6	1.379 (5)	C10—H10	0.9300
C1—C2	1.384 (5)	C11—C12	1.371 (7)
C1—C7	1.493 (5)	C12—C13	1.353 (6)
C2—C3	1.384 (5)	C12—H12	0.9300
C2—H2A	0.9300	C13—C14	1.376 (5)
C3—C4	1.362 (6)	C13—H13	0.9300
C4—C5	1.384 (6)	C14—H14	0.9300
O2—N1—O1	123.8 (4)	O3—C7—N2	123.6 (3)
O2—N1—C3	118.4 (4)	O3—C7—C1	120.3 (3)
O1—N1—C3	117.8 (4)	N2—C7—C1	116.0 (3)
C7—N2—N3	118.5 (3)	N3—C8—C9	120.6 (3)
C7—N2—H2	126 (3)	N3—C8—H8	119.7
N3—N2—H2	115 (3)	C9—C8—H8	119.7
C8—N3—N2	115.5 (3)	C10—C9—C14	119.2 (4)
C6—C1—C2	119.8 (3)	C10—C9—C8	119.4 (3)
C6—C1—C7	118.6 (3)	C14—C9—C8	121.4 (3)
C2—C1—C7	121.5 (3)	C11—C10—C9	118.4 (4)
C3—C2—C1	118.3 (4)	C11—C10—H10	120.8
C3—C2—H2A	120.9	C9—C10—H10	120.8
C1—C2—H2A	120.9	F1—C11—C10	118.9 (5)
C4—C3—C2	122.5 (4)	F1—C11—C12	118.1 (4)
C4—C3—N1	119.0 (4)	C10—C11—C12	122.9 (4)
C2—C3—N1	118.5 (4)	C13—C12—C11	118.2 (4)
C3—C4—C5	118.4 (4)	C13—C12—H12	120.9
C3—C4—H4	120.8	C11—C12—H12	120.9
C5—C4—H4	120.8	C12—C13—C14	121.6 (4)
C6—C5—C4	120.4 (4)	C12—C13—H13	119.2
C6—C5—H5	119.8	C14—C13—H13	119.2
C4—C5—H5	119.8	C13—C14—C9	119.7 (4)
C5—C6—C1	120.6 (4)	C13—C14—H14	120.2
C5—C6—H6	119.7	C9—C14—H14	120.2
C1—C6—H6	119.7		

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N2—H2···O3^{i}	0.90 (1)	1.96 (1)	2.846 (4)	172 (5)

Symmetry code: (i) x, $-y+1/2$, $z-1/2$.